

# Online laser monitoring of metal chloride and oxygen concentration using collinear photofragmentation and atomic absorption spectroscopy

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Collinear photofragmentation and atomic absorption spectroscopy (CPFAAS) is a recently developed optical method for analysing molecular concentrations and their kinetics. [1,2] The technique is based on the fragmentation of a precursor molecule and the detection of the fragment atoms via absorption spectroscopy on a common laser beam path. It has been applied to monitor concentrations of the precursors KCl and KOH in thermal conversion applications. CPFAAS studies have focused on precursor molecule concentration analysis using the maximum absorbance right after the photofragmentation. Sample lengths ranging from 1 cm to 10 m have been demonstrated in the studies [2,3]. In this work, an overview and recent development of CPFAAS is presented.

Typical CPFAAS measurement arrangement is shown in Fig 1. CPFAAS technique has been demonstrated for online monitoring of KCl, KOH, and O<sub>2</sub> in a single particle reactor [2,3] and for KCl and PbCl<sub>2</sub> in full-scale power plant. Typical molecular detection limits reach sub-ppm levels. The spatial resolution of the measurement can be adjusted by tuning the overlapping of the fragmenting and probing laser beams. This allows collection of localized information that is often required to support modelling work of flames and combustion processes. Online measurement of PbCl<sub>2</sub> is a powerful demonstration of the CPFAAS technique's ability to be applied also to monitoring of more complex molecules, such as ZnCl<sub>2</sub>, that are present in combustion processes. In addition, recent development in laser sources has brought NaCl and NaOH monitoring closer to realization.

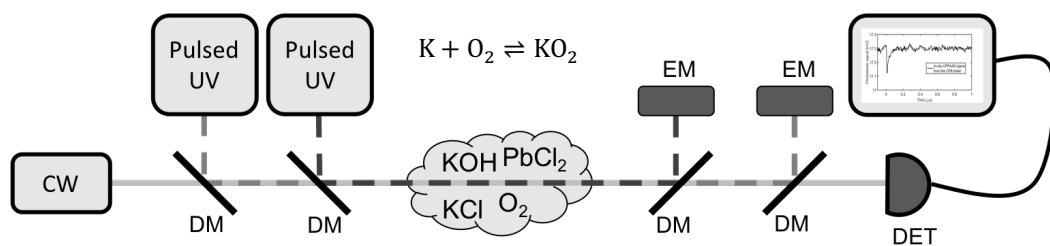


Figure 1. Schematic presentation of typical CPFAAS measurement arrangement showing the laser beam path.

## References

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- [3] J. Viljanen, T. Sorvajärvi, and J. Toivonen, *In situ laser measurement of oxygen concentration and flue gas temperature utilizing chemical reaction kinetics*, Optics letters **42**, 4925-4928 (2017).